

Org. Commun. 10:1 (2017) 24-31



Convenient synthesis and characterization of some novel benzothiazolone-based Schiff bases as potential pharmaceutically active agents

Hanane Chabane, Yasmina Adjeroud and Messaoud Liacha*

Laboratory of Synthesis and Organic Biocatalysis (LSBO), Organic Synthesis and Medicinal Chemistry Group, Department of Chemistry, faculty of Sciences, BADJI Mokhtar-Annaba University, P.O. Box. 12, Annaba 23000, Algeria

(Received May 28, 2016; Revised December 4, 2016; Accepted December 28, 2016)

Abstract: A novel series of Schiff bases derivatives containing the benzothiazolone moiety have been synthesized in a simple and efficient method, by condensation of 6-amino-2(3H)-benzothiazolone substrates with different aromatic aldehydes, in refluxing ethanol and in the presence of acetic acid as catalyst. The structure of synthesized compounds was elucidated and has been proven using spectral methods such as ¹H NMR, ¹³C NMR and elemental analysis. All the newly synthesized compounds were in good agreement with the proposed structures.

Keywords: Schiff bases; 2(3H)-benzothiazolone; 6-amino-2(3H)-benzothiazolones; aromatic aldehydes. © 2017 ACG Publications. All rights reserved.

1. Introduction

Organic compounds with a general formula R¹R²C=NR³ are known as Schiff bases, which are usually synthesized from the condensation of primary amines with compounds having active carbonyl groups,¹⁻³ in different conditions and in different solvents with the elimination of water molecule. Schiff bases are used as a key intermediate for the synthesis of nitrogen heterocyclic compounds.⁴⁻⁶ They play important roles in both synthetic and structural research because of their preparative simplicity and structural diversity.⁷ Several synthetic methods have been reported for their synthesis in the literature.^{8,9} Moreover, the Schiff base derivatives have been extensively studied because of their numerous applications in various fields of chemistry and industry.¹⁰⁻¹⁴ Also, they are reported to possess diverse pharmacological activities.¹⁵⁻²¹

In addition, benzothiazolone containing heterocycles systems have been incorporated into a wide variety of therapeutically interesting drug candidates²². Previous works were published, expanding the structure activity relationship of 6-benzoyl-2(3H)-benzothiazolone (S-14080) as an analgesic compound.²³ 6-benzoyl-2(3H)-benzothiazolone and 6-benzoyl-2(3H)-benzoxazolone also served as lead structures in the design of antiviral compounds, particularly targeted against HIV and CMV species.²⁴ Furthermore, In recent years, several potentially useful bioactive substances based on the benzothiazolone nucleus have been extensively studied because of their wide range of pharmacological activities.²⁵⁻²⁹, and were reported as potential analgesic and anti-inflammatory agents.³⁰⁻³⁴

The article was published by Academy of Chemistry of Globe Publications www.acgpubs.org/OC/index.htm @ Published 01/31/2017 EISSN:1307-6175

^{*} Corresponding author: E-Mail: <u>m_liacha@yahoo.fr</u>

All the facts discussed above and in continuation of our research on nitrogen and sulfur containing heterocycles, inspired us to develop an efficient and simple synthesis of a new series of benzothiazolone derived Schiff's base derivatives (Figure 1), which may exert potent pharmacological action.



 $R=H, CH_3$

Figure 1. Proposed chemical structure of the Schiff base ligands 5a-l

The present investigation is of interest in synthetic organic chemistry because it is based on the fact that, the chemistry of benzothiazolinonyl Schiff bases derivatives is less explored compared to that of either Schiff bases or benzothiazolone.

2. Experimental

Melting points have been determined using a SMP3 Stuart Scientific apparatus and are uncorrected. The ¹H NMR spectra were performed in solutions on a Bruker AC 400 spectrometer using dimethylsulfoxide-d₆ as solvent with TMS as internal standard, with chemical shifts reported as δ (*ppm*). Analytical thin layer chromatography was performed with commercial silica gel plates 60 F₂₅₄ (Merck) and visualized with UV light, using Ethylacetate/Cyclohexane (8:2, v/v) solvent system as eluent. The experimental microanalyses were in satisfactory agreement and were found within 0.4% of the calculated values. The identity of the known products 2(3*H*)-benzothiazolone derivatives **1**, **2**, **3a**, **3b**, **4a** and **4b** was confirmed by the comparison of their melting points and spectroscopic data with those of authentic compounds available in the literature.³⁵⁻⁴⁰

General procedure for the preparation of nitro compounds (*3a-3b*)*:*

Nitric acid (68%, 5.30 cm³, 80 mmol) in 20 cm³ of acetic anhydride cooled to -0.5° C was added dropwise, a solution of 2(3*H*)-benzothiazolone compounds **1** and **2** (10 mmol) in a minimum of acetic anhydride. The mixture was stirred at -0.5° C for 3 h. The precipitate was filtered, washed with cold water, dried and recrystallized from suitable solvent to afford the corresponding 6-nitrobenzothiazolones compounds **3a** (56%) and **3b** (68%).

General procedure for the preparation of amino compounds (4a-4b):

To a stirring ethanolic solution of 6-nitro-2(3*H*)-benzothiazolone (4a) or 3-methyl-6-nitro-2(3*H*)-benzothiazolone (4b) (1.0 equiv.) in a 250 ml round bottomed flask, tin chloride dihydrate (SnCl₂.2H₂O, 5 equiv.) was added. The reaction mixture was heated at reflux and reaction continued until completion of the reaction (TLC monitoring). After complete reduction, the starting material disappeared, and the solution was allowed to cool down. The p*H* was made slightly basic (p*H* 7–8) by addition of 5% aqueous sodium bicarbonate before extraction with ethyl acetate. The organic phase was washed with brine and dried over magnesium sulfate, and the solvent was removed. The solid 6-aminobenzothiazolone intermediates 4a (68%) and 4b (67%), were obtained after being washed with petroleum ether, and used for the next step without further purification.

General procedure for the preparation of Schiff bases ligands (5a-5l):

The Schiff base ligands were prepared by condensation of an equimolar mixture of substituted aromatic aldehydes and 6-amino-2(3H)-benzothiazolones (4a-4b), in an ethanolic solution and in presence of catalytic amount of glacial acetic acid under nitrogen atmosphere. Then, the resulting mixture was heated at reflux for 0.5h until the completion of the reaction (TLC monitoring). The solid product thus obtained was filtered, washed several times with ethanol and cold water, dried and purified to give the desired products (5a-5l). All prepared products were obtained following the general procedure under heat at reflux for 0.5h, as coloured solid products with yield (64-90%). On the

basis of various analytical and NMR spectroscopic data which are given in the experimental section, the structure for the compounds has been proposed (Figure 1).

Physicochemical and spectral data of new products: 6-(Arylideneamino)benzo[d]thiazol-2(3H)-one derivatives (**5a-5l**):

6-(2-cyanobenzylideneamino)benzo[d]thiazol-2(3H)-one (5a): Yellow powder (75%); m.p. 226-227°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 7.19 (d, J = 8.4 Hz, 1H), 7.33-7.35 (dd, $J_1 = 6.4$ Hz, $J_2 = 2.0$ Hz, 1H), 7.70-7.72 (m, 2H), 7.85 (t, $J_1 = 8.0$ Hz, $J_2 = 8.0$ Hz, 1H), 7.99 (d, J = 7.6 Hz, 1H), 8.18 (d, J = 8.0 Hz, 1H), 8.82 (s, N=CH, azomethine, 1H), 12.00 (s, N-H, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 170.0 (C=O), 155.1 (N=C), 145.1, 137.4, 135.6, 134.1, 133.4, 131.5, 128.6, 124.4, 120.7, 117.2, 115.34, 112.0, 111.4 . Anal. Calcd for C₁₅H₉N₃OS: C, 64.50; H, 3.25; N, 15.04; S, 11.48. Found: C, 64.44; H, 3.23; N, 15.90; S, 11.39.

3-methyl-6-(2-cyanobenzylideneamino)benzo[d]thiazol-2(3H)-one (*5b*): Yellow powder (69%); m.p. 189-190°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.44 (s, N-CH₃, 3H), 7.40 (d, *J* = 8.8 Hz, 1H), 7.43-7.46 (dd, *J*₁ = 2.0, *J*₂ = 6.4, 1H), 7.71-7.88 (m, 3H), 7.98-8.01 (dd, *J*₁ = 1.2, *J*₂ = 6.8 Hz, 1H), 8.20 (d, *J* = 7.6 Hz, 1H), 8.84 (s, N=CH, azomethine, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 168.70 (C=O), 155.5 (N=C), 145.7, 137.3, 136.7, 134.1, 133.4, 131.6, 128.7, 122.3, 120.7, 117.2, 115.4, 111.8, 111.5 . 111.8, 29.17 (CH₃). Anal. Calcd for C₁₆H₁₁N₃OS: C, 65.51; H, 3.78; N, 14.34; S, 10.93. Found: C, 64.65; H, 3.23; N, 14.30; S, 10.94.

6-(4-cyanobenzylideneamino)benzo[d]thiazol-2(3H)-one (5c): Yellow powder (90%); m.p. 311-312°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 7.17 (d, J = 8.4 Hz, 1H), 7.32-7.35 (dd, $J_1 = 8.4$ Hz, $J_2 = 2.4$ Hz, 1H), 7.67 (d, J = 2.0 Hz, 1H), 7.98 (d, J = 8.4 Hz, 2H), 8.09 (d, J = 8.4 Hz, 2H), 8.79 (s, N=CH, azomethine, 1H), 12.00 (s, N-H, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 169.9 (C=O), 158.0 (N=C), 145.4, 139.9, 135.3, 133.1, 132.7, 129.8, 128.9, 124.3, 120.7, 118.5, 115.3, 111.9, 113.1 . Anal. Calcd for C₁₅H₉N₃OS: C, 64.50; H, 3.25; N, 15.04; S, 11.48%. Found: C, 64.22; H, 3.23; N, 15.02; S, 11.35%.

3-methyl-6-(4-cyanobenzylideneamino)benzo[d]thiazol-2(3H)-one (5d): Yellow powder (81%); m.p. 205-206°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.43 (s, N-CH₃, 3H), 7.38 (d, *J* = 8.8 Hz, 1H), 7.42-7.45 (dd, *J* = 8.8 Hz, *J* = 2.0 Hz, 1H), 7.74 (d, *J* = 2.0 Hz, 1H), 7.97 (d, *J* = 8.0 Hz, 1H), 7.99 (d, *J* = 8.0 Hz, 1H), 8.08 (s, 1H), 8.10 (s, 1H), 8.81 (s, N=CH, azomethine, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 163.8 (C=O), 157.3 (N=C), 144.6, 140.1, 135.7, 132.7, 131.9, 129.8, 127.4, 122.2, 120.6, 118.6, 115.5, 111.8, 112.0, 29.2 (CH₃). Anal. Calcd for C₁₆H₁₁N₃OS: C, 65.51; H, 3.78; N, 14.34; S, 10.93%. Found: C, 65.62; H, 3.71; N, 14.72; S, 10.18%.

6-(2,4-dimethoxybenzylideneamino)benzo[d]thiazol-2(3H)-one (5e): Yellow powder (66%); m.p. 227-228°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.85 (s, 3H, O-CH₃), 3.89 (s, 3H, O-CH₃), 6.65 (t, $J_1 = 8.8$ Hz, $J_2 = 2.0$ Hz, 1H), 6.67 (d, J = 2.4 Hz, 1H), 7.11 (d, J = 8.4 Hz, 1H), 7.14-7.17 (dd, J = 6.4 Hz, J = 2.0 Hz, 1H), 7.49 (d, J = 2.0 Hz, 1H), 7.96 (d, J = 8.8 Hz, 1H), 8.75 (s, 1H, N=CH, azomethine), 11.86 (s, 1H, N-H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 170.0 (C=O), 163.6 (N=C), 160.7, 153.9, 147.3, 134.1, 128.1, 124.2, 120.1, 117.1, 114.7, 111.8, 106.5, 98.0, 55.8 (OCH₃), 55.5 (OCH₃). Anal. Calcd for C₁₆H₁₄N₂O₃S: C, 61.13; H, 4.49; N, 8.91; S, 10.20%. Found: C, 59.60; H, 4.40; N, 8.76; S, 9.76%.

3-methyl-6-(2,4-dimethoxybenzylideneamino)benzo[d]thiazol-2(3H)-one) (5f): Yellow powder (73%); m.p. 177-178°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.41 (s, N-CH₃, 3H), 3.87 (s, 3H, O-CH₃), 3.89 (s, 3H, O-CH₃), 6.64-6.68 (m, 2H), 7.24-7.27 (dd, $J_1 = 2.0, J_2 = 8.4$ Hz, 1H), 7.31 (d, J = 8.4 Hz, 1H), 7.59 (d, J = 2.0 Hz, 1H), 7.97 (d, J = 8.4 Hz, 1H), 8.779 (s, N=CH, azomethine, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 168.6 (C=O), 163.6 (N=C), 160.7, 154.3, 147.9, 135.4, 128.1, 122.1, 120.2, 117.0, 114.7, 111.6, 106.5, 98.0, 55.8 (OCH₃), 55.5 (OCH₃), 29.0 (CH₃). Anal. Calcd for C₁₇H₁₆N₂O₃S: C, 62.18; H, 4.91; N, 8.53; S, 9.76%. Found: C, 62.18; H, 4.91; N, 8.23; S, 9.38%.

27

6-(2,5-dimethoxybenzylideneamino)benzo[d]thiazol-2(3H)-one (**5g**): Yellow powder (72%); m.p. 196-197°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): δ 3.77 (s, 3H, O-CH₃), 3.85 (s, 3H, O-CH₃), 7.11 (m, 2H), 7.13 (d, J = 8.4 Hz, 1H), 7.21 (dd, $J_1 = 6.4$, $J_2 = 2.0$ Hz, 1H), 7.52 (t, $J_1 = 3.2$, $J_2 = 2.0$, 1H), 7.56 (d, J = 2.0 Hz, 1H), 8.84 (s, N=CH, azomethine, 1H), 11.92 (s, N-H, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm) :170.0 (C=O), 154.2 (N=C), 153.7, 153.1, 146.7, 134.6, 124.4, 124.3, 120.3, 119.4, 114.9, 113.6, 111.8, 110.0, 56.2 (OCH₃), 55.4 (OCH₃). Anal. Calcd for C₁₆H₁₄N₂O₃S: C, 61.13; H, 4.49; N, 8.91; S, 10.20%. Found: C, 59.60; H, 4.40; N, 8.76; S, 9.76%.

3-methyl-6-(2,5-dimethoxybenzylideneamino)benzo[d]thiazol-2(3H)-one (**5h**): Yellow powder (83%); m.p. 165-166°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.42 (s, N-CH₃, 3H), 3.78 (s, O-CH₃, 3H), 3.85 (s, O-CH₃, 3H), 7.11 (s, 1H), 7.12 (s, 1H), 7.33 (s, 1H), 7.32 (d, *J* = 1.6 Hz, 2H), 7.53 (t, *J*₁ = 3.2 Hz, *J*₂ = 1.6, 1H), 7.66 (d, *J* = 1.6 Hz, 1H), 8.86 (s, N=CH, azomethine, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 168.6 (C=O), 154.6 (N=C), 153.8, 153.1, 147.2, 135.9, 124.3, 122.2, 120.4, 119.5, 114.9, 113.6, 111.7, 110.0, 56.2 (OCH₃), 55.4 (OCH₃), 29.0 (CH₃). Anal. Calcd for C₁₇H₁₆N₂O₃S: C, 62.18; H, 4.91; N, 8.53; S, 9.76%. Found: C, 62.36; H, 5.01; N, 8.26; S, 9.44%.

6-(2-chloro-6-nitrobenzylideneamino)benzo[d]thiazol-2(3H)-one (**5i**): Yellow powder (65%); m.p. 232-233°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 7.16 (d, J = 8.4 Hz, 1H4), 7.20-7.23 (dd, J = 8.4 Hz, J = 2.0 Hz, 1H), 7.60 (d, J = 2.0 Hz, 1H), 7.73 (t, $J_1 = 8.0$ Hz, $J_2 = 8.0$ Hz, 1H), 7.93 (d, J = 8.0 Hz, 1H), 7.98 (d, J = 8.0 Hz, 1H), 8.88 (s, N=CH, azomethine, 1H), 12.01 (s, N-H, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 170.2 (C=O), 153.2 (N=C), 149.7, 144.8, 135.2, 133.0, 132.3, 124.4, 120.5, 119.3, 119.1, 116.5, 115.3, 112.1. Anal. Calcd for C₁₄H₈ClN₃O₃S: C, 50.38; H, 2.42; N, 12.59; S, 9.61%. Found: C, 50.40; H, 2.40; N, 12.54; S, 9.47%.

3-methyl-6-(2-chloro-6-nitrobenzylideneamino)benzo[d]thiazol-2(3H)-one (5j): Yellow powder (75%); m.p. 202-203°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.43 (s, 3H, N-CH₃), 7.31-7.33 (dd, $J_1 = 6.8$ Hz, $J_2 = 2.0$ Hz, 1H), 7.39 (d, J = 8.4 Hz, 1H), 7.70 (d, J = 1.6 Hz, 1H), 7.75 (t, $J_1 = 8.0$ Hz, $J_2 = 8.0$ Hz, 1H), 7.95 (d, J = 8.0, 1H), 7.99 (d, J = 8.4 Hz, 1H), 8.92 (s, N=CH, azomethine, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 168.7 (C=O), 153.5 (N=C), 152.7, 149.5, 145.1, 137.0, 134.5, 133.9, 133.7, 132.1, 127.7, 123.2, 120.4, 115.3, 111.8, 29.1 (CH₃). Anal. Calcd for C₁₅H₁₀ClN₃O₃S: C, 51.80; H, 2.90; N, 12.08; S, 9.22%. Found: C, 51.55; H, 2.88; N, 12.44; S, 9.03%.

6-(4-(methylthio)benzylideneamino)benzo[d]thiazol-2(3H)-one (**5k**): Yellow powder (64%); m.p. 220-221°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): 3.32 (s, 3H, S-CH₃), 7.14 (d, J = 8.4 Hz, 1H), 7.23-7.26 (dd, $J_1 = 8.4$ Hz, $J_2 = 2.0$ Hz, 1H), 7.37 (s, 1H), 7.39 (s, 1H), 7.58 (d, J = 2.4 Hz, 1H), 7.83 (s, 1H), 859 (s, 1H), 8.61 (s, N=CH, azomethine, 1H), 11.92 (s, N-H, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 169.9 (C=O), 158.9 (N=C), 146.3, 142.7, 134.5, 132.5, 128.9 (2C), 125.3 (2C), 124.2, 120.4, 114.9, 111.8, 14.1 (S-CH₃). Anal. Calcd for C₁₅H₁₂N₂OS₂: C, 59.97; H, 4.03; N, 9.33; S, 21.35%. Found: C, 59.31; H, 3.99; N, 8.94; S, 21.02%.

3-methyl-6-(4-(methylthio)benzylideneamino)benzo[d]thiazol-2(3H)-one (*5l*): Yellow powder (75%); m.p. 157-158°C. ¹H NMR (400 MHz, DMSO-d₆, δ ppm): δ 5.32 (s, 3H, N-CH₃), 5.43 (s, 3H, S-CH₃), 9.35-9.40 (m, 4H), 9.67 (s, 1H7), 9.85 (d, *J* = 2Hz, 1H), 9.87 (d, *J* = 1.6 Hz, 1H), 10.64 (s, N=CH, azomethine, 1H). ¹³C NMR (100 MHz, DMSO-d₆, δ ppm): 168.5 (C=O), 159.3 (N=C), 146.8, 142.8, 135.8, 132.4, 128.9, 125.3, 122.1, 120.3, 115.1, 111.7, 29.1 (N-CH₃), 14.1 (S-CH₃). Anal. Calcd for C₁₆H₁₄N₂OS₂: C, 61.12; H, 4.49; N, 8.91; S, 20.40. Found: C, 61.01; H, 4.46; N, 9.25; S, 20.36.

3. Results and Discussion

In this work, we have developed a facile and efficient approach for the synthesis of some novel Schiff bases compounds containing the benzothiazolone moiety. The reaction of the starting compounds 6-amino-2(3*H*)-benzothiazolones **4a** and **4b** with substituted benzaldehyde derivatives in ethanol and acetic acid under conventional heating method was studied.

The synthesis of the desired Schiff base 6-(arylideneamino)benzo[d]thiazol-2(3*H*)-one derivatives (**5a-l**) was accomplished according to the steps illustrated in the schemes below. The compound, 3-methyl-2(3*H*)-benzothiazolone (**2**) was prepared by methylation using dimethylsulfate and sodium hydroxide in aqueous medium of 2(3H)-benzothiazolone (**1**), which was readily synthesized via the reaction of 2-aminothiophenol and urea (Scheme 1).



Scheme 1. Reagents and conditions: (a) Δ , 160°C, 3h; (b) dimethylsulfate, NaOH, 2h

The 2(3H)-benzothiazolone (1) and 3-methyl-2(3H)-benzothiazolone (2) were then converted to 6-nitro-2(3H)-benzothiazolone (3a) and 3-methyl-6-nitro-2(3H)-benzothiazolone (3b) derivatives, by a nitration reaction of the aromatic ring of compounds 1 and 2 with nitric acid in acetic anhydride. The nitro group of compounds 3a-3b was reduced by the tin chloride dihydrate (SnCl₂.2H₂O) in refluxing ethanol, yielded the intermediates substrates 6-amino-2(3H)-benzothiazolones 4a and 4b respectively (Scheme 2).



Scheme 2. Reagents and conditions: (c) HNO₃ (68%), acetic anhydride, -5-0°C; (d) SnCl₂.2H₂O, ethanol, reflux, 3h

The final step was performed by condensing 4a and 4b with different aromatic aldehydes at refluxing ethanol, in the presence of catalytic amount of glacial acetic acid. This process yielded 64-90% of the pure desired products (Scheme 3). On the basis of various analytical and NMR spectroscopic data which are given in the experimental section, the structure for the compounds has been proposed (Figure 1).



Scheme 3. Reagents and conditions: (e) appropriate aldehydes, glacial acetic acid, absolute ethanol, reflux, 30 min

All syntheses of the benzothiazolone-based Schiff base derivatives were completed within 30 min. The physical data and yield of the newly synthesized compounds are summarized in the table 1.

4. Conclusion

In summary, this present study reports an efficient strategy and a successful synthesis of a series of novel Schiff base containing benzo[d]thiazol-2(3*H*)-one nucleus in good yields, by reaction of 6-amino-2(3*H*)-benzothiazolone intermediates with various aromatic aldehydes in ethanol and a catalytic amount of glacial acetic acid under reflux conditions. The structural characterizations of the

title compounds were confirmed by using the ¹H NMR, ¹³C NMR, spectral technique and elemental analysis. The employed analytical methods confirmed the structures of the obtained compounds, for both the intermediates and the final products. Finally, it can be concluded that different substituent's on aromatic nucleus influences the activity, and the synthesized Schiff bases products could represent a group of potential agents for the development of new bioactive compounds.

Entry	Product 5	R	R'	Mp. (°C) ^a	Time	Yield	Mol. F.
					(min)	(%) ^b	
1	5a	Н	2-CN	226-227	30	75	C ₁₅ H ₉ N ₃ OS
2	5b	CH ₃	2-CN	189-190	30	69	$C_{16}H_{11}N_3OS$
3	5c	Н	4-CN	311-312	30	90	$C_{15}H_9N_3OS$
4	5d	CH ₃	4-CN	205-206	30	81	$C_{16}H_{11}N_3OS$
5	5e	Н	2,4-OCH ₃	227-228	30	66	$C_{16}H_{14}N_2O_3S$
6	5f	CH ₃	2,4-OCH ₃	177-178	30	73	$C_{17}H_{16}N_2O_3S$
7	5g	Н	2,5-OCH ₃	196-197	30	72	$C_{16}H_{14}N_2O_3S$
8	5h	CH ₃	2,5-OCH ₃	165-166	30	83	$C_{17}H_{16}N_2O_3S$
9	5i	Н	2-Cl,6-NO ₂	232-233	30	65	$C_{14}H_8N_3O_3ClS$
10	5j	CH ₃	2-Cl,6-NO ₂	202-203	30	75	$C_{15}H_{10}N_{3}O_{3}ClS$
11	5k	Н	4-SCH ₃	220-221	30	64	$C_{15}H_{12}N_2OS_2$
12	51	CH ₃	4-SCH ₃	157-158	30	75	$C_{16}H_{14}N_2OS_2$

 Table 1. Physicochemical data of the synthesized compounds 5a-51

^aMelting point; ^bIsolated yield after silica chromatography

Acknowledgements

The authors are thankful to the Algerian Ministry of Higher Education and Scientific Research (MESRS), (Project number: CNEPRU-E01120140064) for the support of this research. The authors would like to thank Pr. G. Kirsch, Laboratory of Molecular Engineering and Pharmacological Biochemistry, Jean Barriol Institute, University of Lorraine, 1 Boulevard Arago, 57070 Metz, France, for carrying out spectral (¹H-NMR, ¹³C-NMR) analysis and microanalysis of the compounds.

References

- [1] Bharti, S. K.; Nath, G.; Tilak R.; Singh, S. K. Synthesis, anti-bacterial and anti-fungal activities of some novel Schiff bases containing 2,4-disubstituted thiazole ring. *Eur. J. Med. Chem.* **2010**, *45*, 651-660.
- [2] Wang, Z.; Gao, J.; Wang, J.; Jin, X.; Zou, M.; Li K.; Kang, P. Spectroscopic analyses on interaction of Amantadine-Salicylaldehyde, Amantadine-5-Chloro-Salicylaldehyde and Amantadine-o-Vanillin Schiff-Bases with bovine serum albumin (BSA). *Spectrochim Acta A. Mol Biomol Spectrosc.* 2011, 83, 511-7.

- [3] da Silva, C. M.; da Silva, D. L.; Modolo, L. V.; Alves, R. B.; de Resende, M. A.; Martins C. V. B.; De Fatima, A. Schiff bases: A short review of their antimicrobial activities. *J. Adv. Res.* **2011**, *2*, 1-8.
- [4] Sinha D.; Tiwari, A. K.; Singh, S.; Shukla, G.; Mishra, P.; Chandra H.; Mishra, A. K. Synthesis, characterization and biological activity of Schiff base analogues of indole-3-carboxaldehyde. *Eur. J. Med. Chem.* **2008**, *43*, 160-165.
- [5] Zhao, X.; Song, D. K.; A. B.; Radbil A. B.; Radbil, A. Schiff bases derived from amines and substituted benzaldehydes exhibit antibacterial, anticancer and antitumour activities. *Russ. J. Appl. Chem.* 2007, *80*, 1373-1375.
- [6] Mirkhani, V.; Moghadam, M.; Tangestanninejad S.; Baltork, M. P.; Rasouli N. Catal. Commun. 2008, 9, 219-223.
- [7] Espinet, P.; Esteruelas, M. A.; Ore, L. A.; Sarrano J. L.; Sola, E. Transition metal liquid crystals: Advanced materials within the reach of the coordination chemist. *Coord. Chem. Rev.* **1992**, *117*, 215-274.
- [8] Schiff, H. Mitteilungen aus dem universitatsl aboratorium in Pisa: Eineneue reihe organischer bas en. (in German). *Justus Liebigs Ann. Chem.* **1864**, *131*, 118-119.
- [9] Shinde, A.; Zangade, S.; Chavan, S.; Vibhute, Y. Microwave induced synthesis of bis-Schiff bases from propane-1, 3-diamine as promising antimicrobial analogs. *Org. Commun.* **2014**, *7*, 60-67.
- [10] Shi, L.; Ge, H.M.; Tan, S.H.; Li, H.Q.; Song, Y.C.; Zhu, H.L.; Tan, R. X. Synthesis and antimicrobial activities of Schiff bases derived from 5-chloro-salicylaldehyde *Eur. J. Med. Chem.* 2007, 42, 558-64.
- [11] Dhar, D. N.; Taploo, C. L. Schiff bases and their applications. J. Sci. Ind. Res. 1982, 41, 501-506; [12] Zhou, Y.; Xu, S.; Guo, L.; Zhang, S.; Lu, H.; Gong, Y.; Gao, F. Evaluating two new synthesized Schiff bases on the corrosion of copper in NaCl solutions, RSC Adv. 2015, 5, 14804-14813.
- [13] Shibuya, Y.; Nabari, K.; Kondo, M.; Yasue, S.; Maeda, K.; Uchida F.; Kawaguchi, H. The copper(II) complex with two didentate schiff base ligands. The unique rearrangement that proceeds under alcohol vapor in the solid state to construct noninclusion structure. *Chem. Lett.* **2008**, *37*, 78-79.
- [14] Roth, A.; Becher, J.; Herrmann, C.; Gorls, H.; Vaughan, G.; Reiher, M.; Klemm D.; Plass, W. Trinuclear copper(II) complexes derived from Schiff-base ligands based on a 6-amino-6-deoxyglucopyranoside: structural and magnetic characterization. *Inorg. Chem.* 2006, 45, 10066-10076.
- [15] Bi S.; Li, G. Characterization and antifungal activity of some transition metal complexes of the schiff base derived from 4-acetylbi-phenyl and S-benzyldithiocarbazate. *Synth. React. Inorg. Met-org Chem.* 1999, 29, 1829-1841.
- [16] Shi, L.; Ge, H. M.; Tan, S. H.; Li, H. Q.; Song, Y. C.; Zhu H. L.; Tan, R. X. Synthesis and antimicrobial activities of schiff bases derived from 5-chloro-salicylaldehyde. *Eur. J. Med. Chem.* 2007, 42, 558–564.
- [17] Gumrukcuoglu, N.; Sokmen, B. B.; Ugras, S.; Ugras H. I.; Yanardag, R. Synthesis, antibacterial, antiurease, and antioxidant activities of some new 1,2,4-triazole schiff base and amine derivatives. *J Enzyme Inhib. Med. Chem.* 2013, 28, 89-94.
- [18] Sztanke, K.; Maziarka, A.; Osinka A.; Sztanke, M. An insight into synthetic schiff bases revealing antiproliferative activities in vitro. *Bioorg. Med. Chem.* **2013**, *21*, 3648-3666.
- [19] Mohana K. N.; Mallesha, L. Synthesis and antiproliferative activity of some new fluorinated schiff bases derived from 1,2,4-triazoles. J. Fluor. Chem. 2013, 156, 15-20.
- [20] More, P. G.; Bhalvankar, R. B.; Pattar, S. C. Synthesis and biological activities of schiff bases of Aminothiazoles. J. Indian Chem. Soc. 2001, 78, 474-475.
- [21] Karakaya, I.; Karabuga, S.; Ulukanli, Z.; Ulukanli, S. Synthesis and antifungal evaluation of imines derived from 3-Amino-2-isopropyl-3H-quinazolin-4-one. *Org. Commun.* **2013**, *6*, 139-147.
- [22] Takashima, T.; Kadoh Y.; Kumada, S. Pharmacological investigations of benzothiazoline derivatives. *Arzneim. Forsch.* **1972**, *22*, 711-715.
- [23] Ferreira, S. H.; Lorenzetti, B.; Devissaguet, M.; Lesiuer, D.; Tsouderos Y. S14080, a peripheral analgesic acting by release of an endogenous circulating opioid-like substance. Br. J. Pharmacol. 1995, 114, 303-308.
- [24] Van derpoorten, K.; Ucar, H.; Andrei, G.; Snoeck, R.; Balzarini, J.; De Clercq E. Synthesis and antiviral activity of 6-benzoyl-benzoxazolin-2-one and 6-benzoyl-benzothiazolin-2-one derivatives. *Antivir. Chem. Chemother.* 1999, 10, 87.
- [25] Yekini, I.; Hammoudi, F.; Martin-Nizard, F.; Yous, S.; Lebegue, N.; Berthelot, P.; Carato, P. Antioxidant activity of benzoxazolinonic and benzothiazolinonic derivatives in the LDL oxidation model. *Bioorg. Med. Chem.* 2009, 17, 7823-7830.
- [26] Ucar, H.; Van derpoorten, K.; Cacciaguerra, S.; Spampinato, S.; Stables, J. P.; Depovere, P.; Isa, M.; Masereel, B.; Delarge, J.; Poupaert, J. H. Synthesis and anticonvulsant activity of 2(3*H*)-benzoxazolone and 2(3*H*)-benzothiazolone derivatives. *J. Med. Chem.* **1998**, *41*, 1138-1145.
- [27] Blanc-Delmas, E.; Lebegue, N.; Wallez, V.; Leclerc, V.; Carmona, C.; Staels, B.; Penicaud, L.; Casteilla, L.; Lonchampt, M.; Dacquet, C.; Chavatte, P.; Berthelot, P.; Lesieur, D. Novel 1,3-dicarbonyl compounds

having 2(3H)-benzazolonic heterocycles as PPARgamma agonists. *Bioorg. Med. Chem.* 2006, 14, 7377-7391.

- [28] Weng, J. Q.; Liu, X. H.; Huang, H.; Tan C. X.; Chen, J. Synthesis, Structure and Antifungal Activity of New 3-[(5-Aryl-1,3,4-oxadiazol-2-yl)methyl]benzo[d]thiazol-2(3*H*)-ones. *Molecules*, **2012**, *17*, 989-1001.
- [29] Önkol, T.; Çakir, B.; Ito, S.; Özçlik, B.; Şahin, M.F. Synthesis of some (5-chloro-2(3*H*)-benzothiazolinone-3-yl)aceto/propanohydrazides towards antimicrobial and antiviral activity. *Turk J Pharm Sci.* 2009, 6, 195-206.
- [30] Abdelazeem, A. H.; Khan, S. I.; White, S. W.; Sufka, K. J; McCurdy, C. R. Design, synthesis and biological evaluation of bivalent benzoxazolone and benzothiazolone ligands as potential antiinflammatory/analgesic agents. *Bioorg Med Chem. Bioog. Med. Chem.* 2015, 23, 3248-3259.
- [31] Dundar, Y.; Çakir, B.; Kupeli, E.; Şahin, M. F. Synthesis of 2(3H)-benzothiazolinone derivatives as analgesic and anti-inflammatory agents. *Turkish J. Pharm. Sci.* 2006, *3*, 51-60.
- [32] Unlu, S.; Onkol, T.; Dundar, Y.; Okcelik, B.; Kupeli, E.; Yesilada, E.; Noyanalpan, N.; Sahin, M. F. Synthesis and analgesic and anti-inflammatory activity of some new 6-acyl-2-benzoxazolinone and 6-acyl-2-benzothiazolinone derivatives with acetic acid and propanoic acid residues. *Arch Pharm Pharm Med Chem.* 2003, 336, 353-361.
- [33] Petrov, O.; Antonova, A.; Kalcheva, V.; Daleva, L. Some N-Mannich bases of 2(3H)-benzothiazolones and their analgesic activity. *Med. Chem. Res.* 1995, 5, 442-448.
- [34] Onkol, T.; Banoglu, E.; Dundar, Y.; Kupeli, E.; Sahin, M. F. Amide derivatives of [6-acyl-2benzothiazolinon-3-yl] acetic acids as potential analgesic and anti-inflammatory compounds. *Med. Chem. Res.* **2010**, *19*, 11-24.
- [35] D'Amico, J. J.; Fuhrhop, R. W.; Bollinger, F. G.; Dahl, W. E. Synthesis of heterocyclic compounds from oaminobenzenethiol and ammonium thiocarbamate. J. Heterocycl. Chem. 1986, 23, 641-645.
- [36] Li, Y.; Li, C.; Jin, K.; Sun, S.; Zhou, X. Synthesis and Biological Activities of Novel Benzoxazolinone/Benzothiazolone Derivatives Containing Oxadiazole Moiety. *Acta. Chimica. Sinica.* 2012, 70, 151-160.
- [37] Zinner H.; Nimmich, W. Benzazole. XIV. Alkylierung, acylierung und hydroxymethylierung des benzthiazolons. *J. prakt. Chem.* **1961**, *14*, 139-149.
- [38] Diouf, O.; Depreux, P.; Lesieur, D.; Poupaert, J.; Caignard, D. Synthesis and evaluation of new 2-piperazinylbenzothiazoles with high 5-HT1A and 5HT3 affinities. *Eur. J. Med. Chem.* **1995**, *30*, 715-719.
- [39] D'Amico, J. J.; Bellinger, F. G.; Freeman, J. J. Synthesis of 2-oxo and 2-thioxo-3(2*H*)benzothiazoleethanimic acid anhydride with acetic acid and related products. *J. Heterocycl. Chem.* **1988**, 25, 1503-1509.
- [40] Suzuki, N.; Kodoya, S.; Donmori, R. Synthetic chemotherapeutic agents. IV. Synthesis of 3-Substituted Thiazolo [5,4-f] quinoline Derivatives. *Chem Pharm Bull.* **1976**, *24*, 1050-1058.



© 2017 ACG Publications