

## Cholinesterase Inhibiting Activity and A New Piperidine Alkaloid from *Lobelia laxiflora* L. Roots (Campanulaceae)

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**Abstract:** The total alkaloidal fraction of *Lobelia laxiflora* L. roots was tested for cholinesterase inhibiting activity using spectrophotometric method. The IC<sub>50</sub> value of the alkaloidal fraction recorded was close to that of eserine (286.3 µg/mL and 270 µg/mL, respectively). This biologically active alkaloidal fraction was subjected to a phytochemical study to isolate and identify its major constituents. Two piperidine alkaloids, N-methyl-2-(2'-methoxybutyl),6-(2''-hydroxybutyl)-Δ<sup>3</sup>-piperidine (**1**) and N-methyl-2-(2'-hydroxybutyl),6-(2''-hydroxybutyl)-Δ<sup>3</sup>-piperidine (**2**), were isolated. The structures of the two compounds were established based on their spectral data, including MS, <sup>1</sup>H- and <sup>13</sup>C-NMR, COSY, HMQC and HMBC spectral experiments. Compound (**1**) is a new natural compound while compound (**2**) was previously isolated from the aerial parts of the same plant.

**Keywords:** *Lobelia laxiflora*; piperidine alkaloids; cholinesterase inhibiting activity. © 2014 ACG Publications. All rights reserved.

### 1. Plant Source

The whole plant of *Lobelia laxiflora* L. was collected from a nursery at Saft El-laban, Giza, Egypt, in April 2011 and was kindly identified by Dr. Mohamed el Gebaly, National Research Institute, Doki, Giza, Egypt and Madam Treze Labib, head of taxonomist, El-Orman Garden, Giza, Egypt. A voucher specimen was kept in the herbarium of Department of Pharmacognosy, Faculty of Pharmacy, Cairo University, Cairo, Egypt (012.10.32).

### 2. Previous Studies

Neurological disorders like Alzheimer's or Parkinson's diseases represent a public health problem. Thus, the discovery of effective agents for the treatment of these diseases is one of the major challenges in herbal medicine [1,2]. Campanulaceae is a family of about 70 genera and 2000 species, most are tropical or subtropical herbs, a few trees or shrubs. Genus *Lobelia* (200-300 species), belonging to subfamily Lobelioideae which has sometimes considered as a separate family (Lobeliaceae) [3], has a long history of therapeutic usage ranging from respiratory stimulant and

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emetic to tobacco smoking cessation [4,5]. This genus characterized by the presence of piperidine alkaloid, which constitute a large class of compounds having various biological activities [6]. Lobeline appears to be the most biologically active alkaloid of *Lobelia* plants.

The most promising bioactivity of lobeline concerns its use in the treatment of CNS diseases [5]. A previous study showed that lobeline improves memory in rodents, probably due to its involvement in cholinergic mechanisms of neurotransmission [7], which claimed to be the most common cause of Alzheimer's disease [8]. *Lobelia laxiflora* L. is a perennial blooming bush, 1 m in height, spread in Central America. A previous study on the aerial parts of this plant resulted in isolation of three new piperidine alkaloids. The isolated alkaloids showed anti-inflammatory activity [9]. This work was carried out on the alkaloidal fraction of *L. laxiflora* L. roots to evaluate its possible cholinesterase inhibiting activity and to isolate and identify the biologically active alkaloids.

### 3. Present Study

Acetylcholinesterase (Electric-eel EC 3.1.1.7), acetylthiocholine iodide and 5,5'-dithiobis[2-nitrobenzoic-acid] (DTNB) were purchased from Sigma (St. Louis, MO, USA). Buffers and other chemicals were of analytical grade. Acetylcholinesterase inhibiting activity was measured according to a slightly modified spectrophotometric method [10]. Acetylthiocholine iodide was used as substrate, while DTNB was used for the measurement of cholinesterase activity. 100 mM sodium phosphate buffer (pH 8.0, 140  $\mu$ L), DTNB (10  $\mu$ L), test solution (20  $\mu$ L) and acetylcholinesterase (20  $\mu$ L) were mixed and incubated for 15 minutes (25° C). The reaction was then initiated by the addition of acetylthiocholine iodide (10  $\mu$ L). The hydrolysis of acetylthiocholine iodide was monitored by the formation of yellow 5-thio-2-nitrobenzoate anion as the result of the reaction of DTNB with thiocholine, released by the enzymatic hydrolysis of acetylthiocholine iodide at a wavelength of 412 nm (15 min). Sample was dissolved in DMSO. All the reactions were performed in triplicate in 96-well micro title plates and monitored in a *SpectraMax 340* (Molecular Devices, USA) spectrometer. The concentrations of the sample that inhibited the hydrolysis of substrate (acetylthiocholine) by 50% (IC<sub>50</sub>) were determined by monitoring the effect of increasing concentrations of these compounds in the assays on the inhibition values. The IC<sub>50</sub> values were then calculated using a software program (GraphPad Prism, version 5.01, Inc., 2007, San Diego California USA).

The alkaloidal fraction exhibited a potent cholinesterase inhibiting activity, and its recorded IC<sub>50</sub> value was found to be equal to 286.3  $\mu$ g/mL comparing to 270  $\mu$ g/mL for eserine as a reference standard.

Dried powdered roots of *Lobelia laxiflora* L. (160 g) were extracted with a mixture of EtOH (95%)-H<sub>2</sub>O (9:1, v/v) till exhaustion. The combined extracts were evaporated under reduced pressure ( $\leq$  60°C) to give 2 g of brown residue. The residue was acidified with 3% HCl (50 mL) and left overnight at room temperature then filtered. The acidic solution was alkalized with 25% NH<sub>4</sub>OH then extracted with CHCl<sub>3</sub> (4 x 50 mL). The CHCl<sub>3</sub> extract was evaporated under reduced pressure to obtain the alkaloidal fraction (350 mg). This fraction (250 mg) was chromatographed on successive silica gel columns using gradient elution with CHCl<sub>3</sub>:MeOH mixtures to obtain compounds **1** (32 mg) and **2** (15 mg).

Precoated silica gel plates 60 F 254 (E-Merck) were used for TLC using S<sub>1</sub> [CHCl<sub>3</sub>:MeOH:NH<sub>4</sub>OH (9:1:2 v/v/drops)] as solvent system. The chromatograms were visualized by spraying with Dragendorff's reagent.

*Compound 1*: Brown resinous substances;  $R_f = 0.5$  (TLC, S<sub>1</sub>); MS  $m/z$ : 255 [M]<sup>+</sup>; <sup>1</sup>H-NMR (CDCl<sub>3</sub>,  $\delta$  ppm): 0.86 (3H, t,  $J=7.5$  Hz, H-4"), 0.92 (3H, t,  $J=7.5$  Hz, H-4'), 1.21 (2H, m, 3' a and 3" a), 1.45 (4H, m, H-3' b, 3" b, 1' a and 1" a), 1.80 (2H, m, H-1' b and 1" b), 2.08 (2H, m, H-5), 2.62 (3H, s, N-CH<sub>3</sub>), 2.83 (3H, s, OCH<sub>3</sub>), 3.62 (1H, m, H-6), 3.76 (1H, m, H-2"), 3.96 (1H, m, 2'), 4.05 (1H, m, H-2), 5.37 (1H, dd,  $J=1.9, 10$  Hz, H-3), 5.77 (1H, m, H-4). <sup>13</sup>C-NMR (CDCl<sub>3</sub>,  $\delta$  ppm): 9.59 (C-4"), 10.2 (C-4'), 26.43 (N-CH<sub>3</sub>), 26.64 (C-5), 30.58 (C-3"), 30.88 (C-3'), 38.23 (C-1"), 38.90 (C-1'), 41.86 (OCH<sub>3</sub>), 62.48 (C-6), 69.86 (C-2), 70.23 (C-2'), 71.36 (C-2"), 124.28 (C-3), 125.66 (C-4).

**Compound 2:** Brown resinous substances;  $R_f = 0.45$  (TLC,  $S_1$ ); MS  $m/z$ : 241  $[M]^+$ ;  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ,  $\delta$  ppm): 0.88 (3H, t,  $J=7.5$  Hz, H-4''), 0.90 (3H, t,  $J=7.5$  Hz, H-4'), 1.18 (2H, m, 3' a, H-3'' a), 1.42 (4H, m, H-3' b, 3'' b, 1' a and 1'' a), 1.71 (2H, m, H-1' b and 1'' b), 2.04 (2H, m, H-5), 2.33 (3H, s, N-CH<sub>3</sub>), 3.65 (1H, m, H-6), 3.71 (2H, m, H-2' and H-2''), 4.17 (1H, m, H-2), 5.39 (1H, dd,  $J=1.9, 10$  Hz, H-3), 5.75 (1H, m, H-4).  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ): 9.67 (C-4''), 9.71 (C-4'), 23.68 (N-CH<sub>3</sub>), 25.79 (C-5), 30.72 (C-3''), 30.91 (C-3'), 37.95 (C-1''), 38.66 (C-1'), 60.19 (C-6), 68.08 (C-2), 71.14 (C-2''), 72.31 (C-2'), 125.55 (C-4), 126.19 (C-3).

Spectral analysis was carried out in Institute of Organic Chemistry, University of Gottingen, Germany. Mass spectrum was operated on LTQ-Orbitrap Velos spectrometer. NMR analysis was performed using  $^1\text{H-NMR}$  (300 MHz),  $^{13}\text{C-NMR}$  (75 MHz): Bruker Avance spectrophotometer.

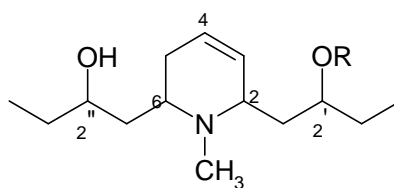
Mass spectrum of compound **1** showed peaks at  $m/z$  255, corresponding to molecular formula  $\text{C}_{15}\text{H}_{29}\text{NO}_2$  and two fragments at 182  $[\text{M}-\text{C}_4\text{H}_9\text{O}]^+$  and 168  $[\text{M}-\text{C}_5\text{H}_{11}\text{O}]^+$  obtained after cleavage of one of the two side chains. A fragment at  $m/z$  96 is typical for N-methylated- $\Delta^3$ -piperidine. NMR spectrum showed signals at  $\delta_{\text{H}}$  3.96 and 3.76 each (1H, m) and at  $\delta_{\text{C}}$  71.36 and 70.23 indicate the presence of two oxygenated tertiary carbons. Singlet at  $\delta_{\text{H}}$  2.83 (3H) assigned for a methyl of a methoxy group. According to HSQC this methoxy group was located on a carbon at  $\delta_{\text{C}}$  41.86. Signals at  $\delta_{\text{C}}$  69.86 and 62.48 corresponding to two carbons neighbored to N atom (C-2 and C-6). Their corresponding protons were displayed at  $\delta_{\text{H}}$  4.17 and 3.65, respectively. Singlet at  $\delta_{\text{H}}$  2.62 (3H) corresponding to N-CH<sub>3</sub>, which located on carbon at  $\delta_{\text{C}}$  26.43 according to HSQC.  $\Delta^3$  is represented by signals of two olefinic protons at  $\delta_{\text{H}}$  5.77 and 5.37 and their respective carbons at  $\delta_{\text{C}}$  124.28 and 125.66. By comparing the spectral data of compound **1** with the published data of similar compounds [9,11,12], it was identified as N-methyl-2(2'-methoxybutyl), 6(2''-hydroxybutyl)- $\Delta^3$ -piperidine, which is a new natural compound. The identification was confirmed by 2D-NMR spectral analysis including HSQC, HMBC and COSY.

Compound **2** showed  $[M]^+$  at  $m/z$  241 corresponding to molecular formula  $\text{C}_{14}\text{H}_{27}\text{NO}_2$  and is typical to  $[M]^+$  of compound **1** after removal of CH<sub>3</sub> of the methoxy group. NMR spectral data of compound **2** showed the absence of the signals at  $\delta_{\text{H}}$  2.83 (OCH<sub>3</sub>) and  $\delta_{\text{C}}$  41.86 (OCH<sub>3</sub>). Thus this compound was identified as N-methyl-2(2'-hydroxybutyl), 6(2''-hydroxybutyl)- $\Delta^3$ -piperidine. This compound was previously isolated from the aerial parts of the same plant [9].

The alkaloidal fraction of *Lobelia laxiflora* L. roots showed potent cholinesterase inhibiting activity comparing to eserine. Previous studies recorded similar activity for lobeline [7]. Thus, this observed bioactivity could be attributed to the piperidine alkaloid content of the root, which was isolated during this course of study. This result suggests the use of *Lobelia laxiflora* L. as a herbal remedy for treatment of neurological disorders like Alzheimer's disease. Further pharmacological and toxicological studies are recommended to establish this finding.

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<b>Compound</b>	<b>1</b>	<b>2</b>
<b>R</b>	<b>CH<sub>3</sub></b>	<b>H</b>

**Figure 1.** Structures of Compounds 1 and 2

## Supporting Information

Supporting Information accompanies this paper on <http://www.acgpubs.org/RNP>

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