

Rec. Nat. Prod. 16:4 (2022) 404-408

records of natural products

# A New Isoflavan Glucoside from the Roots of Astragalus

## membranaceus var. mongholicus

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(Received September 09, 2021; Revised October 18, 2021; Accepted October 24, 2021)

**Abstract:** A new isoflavan glucoside, namely astramemside A (1), together with six known compounds (2–7) were obtained from the roots of a traditional Chinese medicine, *Astragalus membranaceus* var. *mongholicus*. Their structures were elucidated by spectroscopic analyses (HRESIMS, UV, IR, 1D and 2D NMR), and the absolute configuration of 1 was determined by combination of chemical transformation and single-crystal X-ray diffraction. Compound 1 showed moderate inhibition on nitric oxide (NO) production induced by lipopolysaccharide in RAW264.7 cells with an IC<sub>50</sub> value of  $38.98 \pm 5.28 \,\mu$ M.

Keywords: Astragalus membranaceus var. mongholicus; Leguminosae; isoflavan glucoside. © 2021 ACG Publications. All rights reserved.

#### **1. Plant Source**

In this phytochemical study of the roots of *Astragalus membranaceus* var. *mongholicus* (Leguminosae) (Plant materials see supporting information), a new isoflavan glucoside, astramemside A (1), together with six known compounds (2-7) were isolated. Herein, we report the isolation and structural elucidation of these compounds.

### 2. Previous Studies

Astragalus membranaceus var. mongholicus, also known as "Huang-Qi" in traditional Chinese medicine (TCM), is a perennial herb widely distributed in North, Northeast, and Northwest China [1]. Its roots have long been used in TCM for various purposes, such as antiperspirant, antimicrobial, antiinflammatory, diuretic and tonic, etc [2]. Particularly, the extract of this plant can enhance immune system of human [3]. Previous studies on chemical composition of this species showed that isoflavonoids and triterpene saponins were major constituents [4-10]. Some of these constituents

The article was published by ACG Publications

http://www.acgpubs.org/journal/records-of-natural-products July-August 2022 EISSN:1307-6167 DOI: http://doi.org/10.25135/mp.292.2109.2203

Available online: November 8, 2021

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exhibited diverse pharmacological properties such as triglyceride accumulation inhibitory and antiinflammatory activities [4,11].

#### **3. Present Study**

The EtOAc fraction of *A. membranaceus var. mongholicus* was separated by repeated column chromatography over silica gel, reversed phase  $C_{18}$  (RP- $C_{18}$ ), LH-20 gel, and finally HPLC to obtain compounds **1–7** (Figure 1) (detailed separation process see supporting information).

Astramemside A (1): white amorphous powder,  $[\alpha]^{25_{\rm D}}$  -30.8 (*c* 0.2, MeCN); UV (MeCN)  $\lambda_{\rm max}$  (log  $\varepsilon$ ) 280 (2.87), 206 (4.11) nm; IR (KBr)  $\nu_{\rm max}$  3395, 2921, 2850, 1712, 1619, 1505, 1464, 1171, 1095 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; HRESIMS *m*/*z* 555.1838 [M + Na]<sup>+</sup> (calcd for C<sub>27</sub>H<sub>32</sub>O<sub>11</sub>Na<sup>+</sup>, 555.1837).

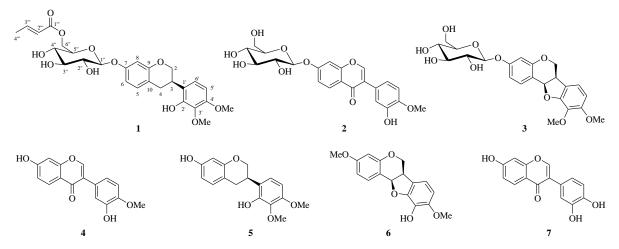


Figure 1. Structures of compounds 1-7

Compound 1, white amorphous powder, displayed a negative optical rotation ( $[\alpha]^{25_{\rm D}}$  -30.8, in MeCN). Its molecular formula,  $C_{27}H_{32}O_{11}$ , was established by the HRESIMS ion peak at m/z 555.1838  $[M + Na]^+$  (calcd for C<sub>27</sub>H<sub>32</sub>O<sub>11</sub>Na<sup>+</sup>, 555.1837). In the IR spectrum, the absorption bands of hydroxy  $(3395 \text{ cm}^{-1})$ , carbonyl group  $(1712 \text{ cm}^{-1})$  and aromatic ring  $(1619, 1505 \text{ and } 1464 \text{ cm}^{-1})$  were observed. The <sup>1</sup>H NMR spectroscopic data (Table 1) displayed signals for a 1,3,4-trisubstituted benzene ring [ $\delta_{\rm H}$  6.97 (1H, d, J = 8.3 Hz, H-5), 6.59 (1H, dd, J = 8.3, 2.5 Hz, H-6) and 6.56 (1H, d, J =2.5 Hz, H-8)], a 1,2,3,4-tetrasubstituted benzene ring [ $\delta_{\rm H}$  6.77 (1H, d, J = 8.7 Hz, H-6') and 6.46 (1H, d, J = 8.7 Hz, H-5')], two *trans*-olefinic protons [ $\delta_{\rm H}$  7.03 (1H, dq, J = 15.6, 6.9 Hz, H-3''') and 5.90 (1H, dd, J = 15.6, 1.6 Hz, H-2''')], two methoxy groups [ $\delta_{\rm H}$  3.81 (3H, s, 4'-OMe) and 3.79 (3H, s, 3'-OMe)], two oxygenated methylenes [ $\delta_{\rm H}$  4.51 (1H, dd, J = 11.8, 2.1 Hz, H-6"a), 4.22 (1H, dd, J = 11.8, 7.5 Hz, H-6"b), 4.27 (1H, brd, J = 9.9 Hz, H-2a) and 4.02 (1H, t, J = 9.9 Hz, H-2b)], five oxygenated methines [ $\delta_{H}$  4.81 (1H, d, J = 7.5 Hz, H-1"), 3.66 (1H, m, H-5"), 3.46 (1H, m, H-3"), 3.45 (1H, m, H-2") and 3.35 (1H, m, H-4")], a methyl group [ $\delta_{\rm H}$  1.85 (3H, dd, J = 6.9, 1.6 Hz, Me-4")]. The <sup>13</sup>C NMR spectrum of 1 showed 27 carbon signals, which can be classified by DEPT and HSQC spectra as a carbonyl ( $\delta_{\rm C}$  168.0), two benzene rings ( $\delta_{\rm C}$  158.3, 156.4, 153.2, 149.6, 137.6, 131.1, 122.9, 122.3, 117.8, 110.6, 105.6 and 104.4), a disubstituted double bond ( $\delta_c$  146.9 and 123.2), a hexosyl group ( $\delta_c$ 102.5, 77.9, 75.5, 74.9, 71.9 and 64.7), two methoxy groups ( $\delta_c$  61.6 and 56.3), two sp<sup>3</sup> methylenes (one oxygenated at  $\delta_{\rm C}$  71.1), a sp<sup>3</sup> methine and a methyl. The aforementioned spectroscopic data suggested that compound 1 was an isoflavan glucoside.

The 2D structure of **1** was determined by analysis of its 2D NMR data (Figure S1). The  ${}^{1}H{-}^{1}H$  COSY correlations of H<sub>2</sub>-2/H-3/H<sub>2</sub>-4, H-5/H-6, H-5'/H-6', in combination with the HMBC correlations of H<sub>2</sub>-2/C-9; H<sub>2</sub>-4/C-5, C-9 and C-10; H-5/C-7 and C-9; H-8/C-6; H-5'/C-3'; H-6'/C-3, C-

2' and C-4' demonstrated the existence of 7,2',3',4'-tetrahydroxyisoflavan moiety. The HMBC
correlations from 3'-OMe ( $\delta_{\rm H}$ 3.79) to C-3' and from 4'-OMe ( $\delta_{\rm H}$ 3.81) to C-4' suggested that the
methoxy groups were attached to C-3' and C-4', respectively. The hexosyl group was determined to be
a $\beta$ glucose by comparison of its 1D NMR data and coupling constant ( $J_{1''/2''} = 7.5$ ) with reported data
[12]. This glucosyl was located at C-7 of the isoflavan moiety as indicated by the HMBC correlation
from the anomeric proton ( $\delta_{\rm H}$ 4.81, H-1") to the C-7 ( $\delta_{\rm C}$ 158.3). The <sup>1</sup> H– <sup>1</sup> H COSY correlations of H-
2"'/H-3"'/H <sub>3</sub> -4"' and HMBC correlation of H-3"'/C-1"' suggested the presence of a (E)-but-2-enoyl
group, and this group was linked to 6"-OH by HMBC from H <sub>2</sub> -6" to C-1". Thus, the planar structure
of compound 1 was determined.

The absolute configuration of 1 was confirmed by chemical transformation. The alkaline hydrolysis of 1 yielded a known product, (3R)-(-)-7,2'-dihydroxy-3',4'-dimethylisoflavan-7-O- $\beta$ -D-glucopyranoside (1a), whose absolute configuration was determined by single-crystal X-ray diffraction [with a Flack parameter of 0.05(5)] (Figure 2). Thus, the structure of compound 1 was established as depicted and named astramemside A.

No.	$\delta_{\rm H}(J \text{ in Hz})$	δc	No.	$\delta_{\rm H}(J \text{ in Hz})$	$\delta_{\mathrm{C}}$
2a	4.27, brd (9.9)	71.1	5'	6.46, d (8.7)	104.4
2b	4.02, t (9.9)		6'	6.77, d (8.7)	122.9
3	3.48, m	33.4	1''	4.81, d (7.5)	102.5
4	a 3.00, dd (15.8, 10.4)	31.2	2''	3.45, m	74.9
	b 2.86, dd (15.8, 4.5)		3''	3.46, m	77.9
5	6.97, d (8.3)	131.1	4''	3.35, m	71.9
6	6.59, dd (8.3, 2.5)	110.6	5''	3.66, m	75.5
7		158.3	6′′a	4.51, dd (11.8, 2.1)	64.7
8	6.56, d (2.5)	105.6	6′′b	4.22, dd (11.8, 7.5)	
9		156.4	1′′′		168.0
10		117.8	2'''	5.90, dd (15.6, 1.6)	123.2
1′		122.3	3′′′	7.03, dq (15.6, 6.9)	146.9
2'		149.6	4'''	1.85, dd, (6.9, 1.6)	18.2
3'		137.6	3'-OMe	3.79, s	61.1
4′		153.2	4'-OMe	3.81, s	56.3

Table 1. <sup>1</sup>H (500 MHz) and <sup>13</sup>C (125 MHz) NMR data of 1 ( $\delta$  in ppm) in CD<sub>3</sub>OD

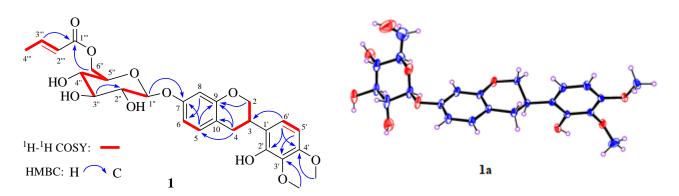


Figure 2. Key <sup>1</sup>H–<sup>1</sup>H COSY and HMBC correlations of 1 (left); X-ray structure of 1a (right)

The known compounds were identified as 7-O- $\beta$ -D-glucopyranosyl 7,3'-dihydroxy-4'-methoxy isoflavone (2) [13], (6a*R*,11a*R*)-9,10-dimethoxypterocarpan-3-O- $\beta$ -D-glucoside (3) [14], calycosin (4) [15], isomucronulatol (5) [16], (6a*R*, 11a*R*)-10-hydroxy-3,9-dimethoxypterocarpan (6) [17], and 3'-hydroxydaidzein (7) [18], by comparison of their spectroscopic data with the published data.

Most of these co-isolated known compounds have been previously reported to possess inhibitory activity on the nitric oxide (NO) production. Among them, compounds **2** and **3** showed strong inhibitory activity, with with IC<sub>50</sub> values of  $4.10 \pm 0.10$  and  $14.70 \pm 0.90 \ \mu$ M, respectively [19]. Compound **4** had moderate inhibitory activity (IC<sub>50</sub> = 39.56 ± 2.43  $\mu$ M), while compound **7** was inactive (IC<sub>50</sub> > 100  $\mu$ M [20]. The new compound **1** was tested for its inhibitory effect on the NO production induced by lipopolysaccharide in RAW264.7 cells. Quercetin was used as a positive control (IC<sub>50</sub> = 17.86 ± 2.13  $\mu$ M). The result showed that compound **1** showed moderate inhibitory activity (IC<sub>50</sub> = 38.98 ± 5.28  $\mu$ M).

#### Acknowledgments

This work was financially supported by the National Key R&D Program of China (grant number 2020YFC1712700).

#### **Supporting Information**

Supporting Information accompanies this paper on <u>http://www.acgpubs.org/journal/records-of-natural-products</u>

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