## Supporting Information

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## Two New Sesquiterpenoids from Kalimeris shimada

Yun-Peng Sun ${ }^{1}$, Yu-Fei Huang ${ }^{1}$, Yang Yu ${ }^{1}$, Jai-Sing Yang ${ }^{2}$, Jin-<br>Song Liu ${ }^{\text {** }}$ and Guo-Kai Wang ${ }^{1,3^{*}}$<br>${ }^{1}$ School of Pharmacy, Anhui University of Chinese Medicine, Hefei 230012, P.R. China<br>${ }^{2}$ Department of Medical Research, China Medical University Hospital, China Medical University, Taichung, Taiwan<br>${ }^{3}$ Anhui Province Key Laboratory of Research \& Development of Chinese Medicine, Hefei 230012, P.R. China

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Elemental Composition Calculator

| Target m/z: | 287.1044 | Result type: | Positive ions | Species: |
| :---: | :---: | :---: | :---: | :---: |
| Elements: | $\mathrm{C}+\mathrm{K}]^{+}$ |  |  |  |
| C (0-80); H (0-120); O (0-30); K (0-5) |  |  |  |  |
| Ion Formula | Calculated m/z |  | PPM Error |  |
| C15H20KO3 | 287.1044 | -0.16 |  |  |

Figure S1: HR-ESI-MS spectrum of 1 (kalshinoid G)


Figure S2: UV spectrum of $\mathbf{1}$


Figure S3: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ spectrum of $\mathbf{1}$ (kalshinoid G)


Figure S4: ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ spectrum of $\mathbf{1}$ (kalshinoid G)


Figure S5: HSQC spectrum of $\mathbf{1}$ (kalshinoid G)


Figure S6: HSQC spectrum of $\mathbf{1}$ (kalshinoid G) (From $\delta_{\mathrm{C}} 24 \mathrm{ppm}$ to $\left.\delta_{\mathrm{C}} 43 \mathrm{ppm}\right)$


Figure S7: HMBC spectrum of $\mathbf{1}$ (kalshinoid G)


Figure S8: HMBC spectrum of $\mathbf{1}$ (kalshinoid G) (From $\delta_{\mathrm{C}} 5 \mathrm{ppm}$ to $\left.\delta_{\mathrm{C}} 90 \mathrm{ppm}\right)$


Figure S9: ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectrum of 1 (kalshinoid G)


Figure S10: ROESY spectrum of $\mathbf{1}(\mathrm{kalshinoid} G)\left(\right.$ From $\delta_{\mathrm{H}} 1.9 \mathrm{ppm}$ to $\left.\delta_{\mathrm{H}} 3.0 \mathrm{ppm}\right)$


Figure S11: ROESY spectrum of $\mathbf{1}$ (kalshinoid G) (From $\delta_{\mathrm{H}} 1.0 \mathrm{ppm}$ to $\left.\delta_{\mathrm{H}} 1.6 \mathrm{ppm}\right)$

Data File: E:IDATAL202210303iwyf-81.Icd

| Elmt | Val. | Min | Max | Elmt | Val. | Min | Max | Elmt | Val. | Min | Max | Elmt | Val. | Min | Max | Use Adduct |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| H | 1 | 5 | 100 | F | 1 | 0 | 0 | Cl | 1 | 0 | 0 | Ag | 1 | 0 | 0 | Na |
| 2 H | 1 | 0 | 0 | Na | 1 | 0 | 0 | Co | 2 | 0 | 0 | I | 3 | 0 | 0 |  |
| B | 3 | 0 | 0 | Mg | 2 | 0 | 0 | Cu | 2 | 0 | 0 | Ir | 3 | 0 | 0 |  |
| C | 4 | 5 | 50 | Si | 4 | 0 | 5 | Se | 2 | 0 | 0 |  |  |  |  |  |
| N | 3 | 0 | 0 | P |  | 0 | 0 | Br | 1 | 0 | 0 |  |  |  |  |  |
| 0 | 2 | 0 | 30 | S | 2 | 0 | 0 | Pd | 2 | 0 | 0 |  |  |  |  |  |
| Error | rgin HC $\times$ Isot Iso R | m): atio: pes: (\%): |  |  |  |  |  |  |  |  |  | Electro Use M Isoto Max | lons Info Res sults: | $\begin{aligned} & \begin{array}{l} \text { both } \\ \text { yes } \\ 10000 \\ 20 \end{array} \end{aligned}$ |  |  |




C15 H26 O3 $[\mathrm{M}+\mathrm{Na}]+$ : Predicted region for $277.1774 \mathrm{~m} / \mathrm{z}$


Figure S12: HR-ESI-MS spectrum of 2 (kalshinoid H)


Figure S13: UV spectrum of 2


Figure S14: ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(600 \mathrm{MHz}\right.$, Acetone- $\left.d_{6}\right)$ spectrum of $2($ kalshinoid H$)$


Figure S15: ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(150 \mathrm{MHz}\right.$, Acetone- $\left.d_{6}\right)$ spectrum of $2($ kalshinoid H$)$


Figure S16: HSQC spectrum of 2 (kalshinoid H)


Figure S17: HSQC spectrum of 2 (kalshinoid H) (From $\delta_{\mathrm{C}} 10 \mathrm{ppm}$ to 50 ppm$)$


Figure S18: HMBC spectrum of 2 (kalshinoid H)


Figure S19: HMBC spectrum of 2 (kalshinoid H) (From $\delta_{\mathrm{C}} 15 \mathrm{ppm}$ to 85 ppm$)$


Figure S20: HMBC spectrum of $\mathbf{2}$ (kalshinoid H) (From $\delta_{\mathrm{C}} 10 \mathrm{ppm}$ to 145 ppm )


Figure S21: ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY spectrum of 2 (kalshinoid H)


Figure S22: ROESY spectrum of $\mathbf{2}$ (kalshinoid H)


Figure S23: ROESY spectrum of $\mathbf{2}$ (kalshinoid H) (From $\delta_{\mathrm{H}} 0.8 \mathrm{ppm}$ to 2.5 ppm$)$


Figure S24. DP4+ analyses of calculated and experimental NMR chemical shifts of 1. Isomer 1:
$1 S^{*}-\mathbf{1}$; Isomer 2 : $1 R^{*}-\mathbf{1}$

Table S1 : Energy analyses of conformers $(1 R, 4 R, 5 S, 10 R)-1 \mathrm{a}-\mathrm{d}$

| NO. | 3D comformers | Free energy |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  | E (Hartree) | $\Delta \mathrm{E}(\mathrm{Kcal} / \mathrm{mol})$ | Boltzmann distribution |
| 1a |  | -808.4255828 | 0.0000 | 53.69\% |


| 1 b | -808.4248818 | 0.000700923 | $25.55 \%$ |  |
| :--- | :--- | :--- | :--- | :--- |
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| 1 c |  | -808.4244607 | 0.704122371 | $16.35 \%$ |
| :--- | :--- | :--- | :--- | :--- |
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Table S2 : Energy analyses of conformers ( $1 S, 4 R, 5 S, 10 R$ )-1a-d

| NO. | 3D comformers |  | Free energy |  |  |
| :--- | :---: | :--- | :--- | :--- | :--- |
|  |  | $\mathrm{E}($ Hartree $)$ | $\Delta \mathrm{E}(\mathrm{Kcal} / \mathrm{mol})$ | Boltzmann distribution |  |
| 2 a |  | -808.4255828 | 0.0000 | $53.69 \%$ |  |
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| :---: | :---: | :---: | :---: | :---: |
| 2b |  | -808.4248818 | 0.000700923 | 25.55\% |


| 2c | -808.4244607 | 0.704122371 | $16.35 \%$ |  |
| :--- | :--- | :--- | :--- | :--- |
|  |  |  |  |  |
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| 2 d | -808.4232252 | 0.002357659 | $4.41 \%$ |  |
| :--- | :--- | :--- | :--- | :--- |
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## NMR Computational details of compound 1.

The initial conformational analysis of the compound $\mathbf{1}$ was executed by employing Monte Carlo searching algorithm via the MMFF94 molecular mechanics force field[1], with the aid of the SPARTAN'16 program package, leading to afford a panel of relatively favored conformations in an energy range of $3 \mathrm{kcal} / \mathrm{mol}$ above the global minimum. The force field minimum energy conformers thus obtained were subsequently optimized by applying the density functional theory (DFT) with the B3LYP/6-31G(d) level in vacuum, implemented in the Gaussian 09 software package[2]. Harmonic vibrational frequencies were also performed to confirm no imaginary frequencies of the finally optimized conformers. Gauge-Independent Atomic Orbital (GIAO) calculations of NMR chemical shifts were accomplished by DFT at the mPW1PW91/6-311+g (d, p) level in Chloroform with the PCM solvent model in Gaussian 09 software. NMR chemical shifts of TMS were calculated in the same level and used as the references. Regression analysis of calculated versus experimental NMR chemical shifts of $\mathbf{1}$ was carried out. Linear correlation coefficients ( $\mathrm{R}^{2}$ ) and Root-mean-square deviation (RMSD) were calculated for the evaluation of the results. After Boltzmann weighing of the predicted chemical shift of each isomers, the DP4+ parameters were calculated using the excel file provided by Ariel M. Sarotti.[3]

## Measurement of NO production

NO production was quantified by measuring the accumulation of nitrite in the cell culture supernatant with Griese reagent [4]. Briefly, RAW 264.7 cells ( $6 \times 10^{6}$ cells $/ \mathrm{mL}$ ) were seeded in 96 -well plates and pretreatment with compounds for 1 h before LPS $(1 \mu \mathrm{~g} / \mathrm{mL})$ stimulation. The isolated culture supernatant was mixed with Griese reagent (Beyotime Biotechnology, China). $\mathrm{NaNO}_{2}$ was used to generate a standard curve, and the absorbance of the mixture was measured at 540 nm . In the experiment, monomethylarginine monoacetate (L-NMMA) was used as a positive control.

## References

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## New compounds search report of SciFinder


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