

## Ionol Derivatives from *Euphorbia tirucalli*

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**Abstract:** A new ionol derivative, (2*R*,6*S*,9*S*)-2-hydroxy-3-oxo- $\alpha$ -ionol (**1**), together with three known analogues, (6*S*,9*S*)-6-hydroxy-3-oxo- $\alpha$ -ionol (**2**), (6*R*,9*S*)-3-oxo- $\alpha$ -ionol  $\beta$ -D-glucopyranoside (**3**) and (6*R*,9*R*)-3-oxo- $\alpha$ -ionol  $\beta$ -D-glucopyranoside (**4**), were isolated from the ethyl acetate extract of the aerial parts of *Euphorbia tirucalli*. Their structures were elucidated by means of extensive spectroscopic methods and comparison with the data reported in the literature. The absolute configuration of **1** was deduced by comparing experimental and calculated ECD spectra and <sup>13</sup>C NMR data. The ionol derivatives have been obtained for the first time from the genus *Euphorbia*.

**Keywords:** *Euphorbia tirucalli*; ionol derivatives; structure elucidation. © 2017 ACG Publications. All rights reserved.

### 1. Introduction

*Euphorbia* is the largest genus of the family Euphorbiaceae, comprising of more than 2000 species in the world and over 80 species in China [1], many of which have been used in folk as traditional Chinese medicine for the treatment of skin diseases, edemas, etc [2]. Previous chemical investigations revealed that diterpenoids, triterpenoids, flavonoids, phenolic acids, tannins and other constituents exist in this genus [3-8]. *Euphorbia tirucalli*, widely cultivated in tropical and subtropical regions, and the north and south areas in China [9], has been used as a traditional medicine in Africa and Asia for purgation and treatment of neuropathic pain, rheumatism, and toothache [10]. Previous researches have shown that the latex of *E. tirucalli* is toxic and strongly excitant on the skin and mucous membrane because of highly unsaturated diterpene ester [11-15]. Our previous research reported the constituents of non-diterpenoids from *E. tirucalli* [16].

As an ongoing program to access chemical diversity of *Euphorbia* and their biological effects, we carried out an investigation on the air-dried of *E. tirucalli*, as a result, a new ionol derivative, (2*R*,6*S*,9*S*)-2-hydroxy-3-oxo- $\alpha$ -ionol (**1**), along with three known analogues, (6*S*,9*S*)-6-hydroxy-3-

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oxo- $\alpha$ -ionol (**2**) [17], (6*R*,9*S*)-3-oxo- $\alpha$ -ionol  $\beta$ -D-glucopyranoside (**3**) [18] and (6*R*,9*R*)-3-oxo- $\alpha$ -ionol  $\beta$ -D-glucopyranoside (**4**) [18], were isolated from the ethyl acetate extract of the aerial parts of this plant. To the best of our knowledge, the ionol derivatives were firstly obtained from the genus *Euphorbia*, which provided new evidences for the chemical diversity of *Euphorbia* plants. In this paper, the isolation and structural elucidation of the new compound are presented.

## 2. Materials and Methods

### 2.1. General procedures

1D and 2D NMR spectra were recorded on a Bruker Avance III-600 and a Bruker AM-400 instruments with TMS as internal standard. UV spectra were carried out on a Shimadzu UV-2401A spectrophotometer. IR spectra were measured on a Bruker Tensor 27 FTIR spectrometer with KBr pellets. Optical rotations were recorded using a Jasco P-1020 Polarimeter. ESI-MS spectra were recorded on a Waters Xevo TQ-S UPLC Triple Quadrupole Mass Spectrometer. HR-ESI-MS data were obtained on an Agilent G6230 Q-TOF mass instrument. Column chromatography (CC) was performed using silica gel (Qingdao Marine Chemical Factory, China, 200–300 mesh), Sephadex LH-20 (Pharmacia Biotech Ltd., Sweden) and MCI gel (CHP 20P, Mitsubishi Corporation, Japan). Thin-layer chromatography (TLC) and preparative TLC were performed using precoated silica gel GF<sub>254</sub> plates (Qingdao Marine Chemical Factory). Semipreparative HPLC was performed on a Hitachi Chromaster system (Hitachi, Ltd., Japan) equipped with an YMC-Triart C<sub>18</sub> column (250 mm  $\times$  10 mm i.d., 5  $\mu$ m, YMC Corporation, Japan), using a flow rate of 3.0 mL/min at a column temperature of 25 °C, and detection was performed with a DAD detector.

### 2.2 Plant Material

The aerial parts of *E. tirucalli* were collected in August 2009 from Xishuang Banna prefecture, Yunnan Province, People's Republic of China, and were identified by Prof. Yao-Wen Yang, Yunnan University of Traditional Chinese Medicine, where a voucher specimen (YTCM 20090803) has been deposited.

### 2.3 Extraction and Isolation

The air-dried and powdered aerial parts of *E. tirucalli* (4.8 kg) were extracted with 70% aqueous acetone (8 L  $\times$  3) at room temperature. The extracts were concentrated by rotary evaporator under reduced pressure to remove organic solvent. The aqueous residue was then partitioned with petroleum ether (4  $\times$  1 L), EtOAc (4  $\times$  1 L), and n-BuOH (4  $\times$  1 L), sequentially. The EtOAc extract (64.0 g) was subjected to MCI gel (CHP 20P) CC using a gradient system of CH<sub>3</sub>OH–H<sub>2</sub>O (30:70, 50:50, 70:70, 90:10) to afford four fractions (Fr A–D).

Fraction B (7.0 g) was chromatographed on Sephadex LH-20 column eluted by MeOH to give two fractions (Fr B-1–2) based on TLC analysis. Fr B-1 (1.2 g) was subjected to column chromatography (CC) on silica gel (200–300 mesh) eluting with CHCl<sub>3</sub>–(CH<sub>3</sub>)<sub>2</sub>CO (30:1–4:1) to afford two fractions (Fr B-1-1–2). Fr B-1-1 (298.7 mg) was purified by Sephadex LH-20 column (MeOH–CHCl<sub>3</sub>, 1:1), followed by semipreparative HPLC (MeOH–H<sub>2</sub>O 27:73) to yield compound **2** (6.5 mg,  $t_R$  = 23.0 min), and by semipreparative HPLC (MeOH–H<sub>2</sub>O 35:65) to yield compound **1** (2.6 mg,  $t_R$  = 29.2 min). Fr B-1-2 (125.4 mg) was chromatographed on Sephadex LH-20 column (MeOH–CHCl<sub>3</sub>, 1:1), followed by semipreparative HPLC (MeOH–H<sub>2</sub>O 40:60) to afford compounds **3** (4.2 mg,  $t_R$  = 23.0 min), and **4** (1.0 mg,  $t_R$  = 29.0 min).

### 3. Results and Discussion

Compound **1**,  $[\alpha]_D^{26.5} +136.2$  (*c* 0.01, MeOH), UV (MeOH)  $\lambda_{\max}$  ( $\log \epsilon$ ): 235 (3.08) nm which revealed the presence of conjugated system, obtained as colorless powder in MeOH. Its molecular formula was determined to be  $C_{13}H_{20}O_3$  based on the HR-ESI-MS data ( $m/z$  223.1337 [M-H], calcd. 223.1340), corresponding to 4 degrees of unsaturation. Its IR spectrum showed absorption bands for hydroxyl group at  $3423\text{ cm}^{-1}$  and conjugated carbonyl group at  $1672\text{ cm}^{-1}$ . The  $^1\text{H}$  NMR spectrum (Table 1) showed three 3H-singlets at  $\delta_H$  1.85, 0.96 and 0.81, and one 3H-doublet at  $\delta_H$  1.11. An olefinic bond proton at  $\delta_H$  5.84 (1H, br s) and two olefinic protons [ $\delta_H$  5.59 (1H, dd,  $J = 15.2, 8.6$  Hz); 5.63 (1H, dd,  $J = 15.2, 5.1$  Hz)] were also evident in the  $^1\text{H}$  NMR spectrum. Additionally, the signals of two oxygenated methines [ $\delta_H$  4.14 (1H, br t,  $J = 5.6$  Hz), 3.95 (1H, s)] were also observed.  $^{13}\text{C}$  NMR spectrum of **1** displayed most upfield resonance due to a conjugated carbonyl at  $\delta_C$  198.7, along with four signals due to olefinic carbons ( $\delta_C$  162.0, s; 139.8, d; 124.3, d; 123.7, d), four methyl carbon resonances including two geminal methyls at  $\delta_C$  24.6 (s), 20.8 (s), one allelic methyl at  $\delta_C$  22.9 (s), and one primary methyl at  $\delta_C$  23.9. Accordingly, **1** was presumably an oxygenated ionol derivative substituted by a hydroxyl group and a carbonyl function.

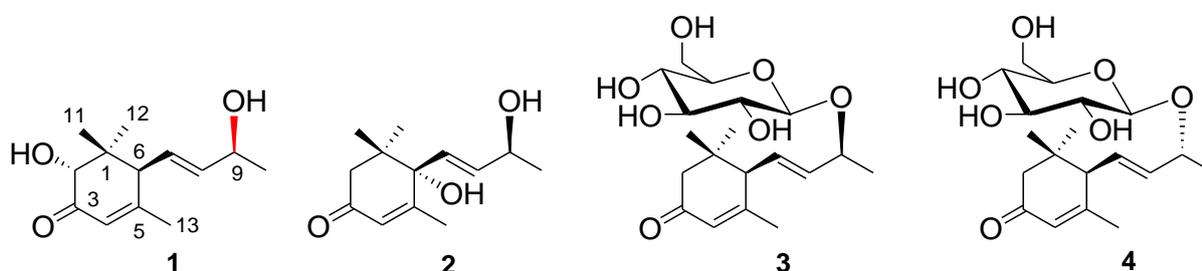


Figure 1. The chemical structures of compounds 1-4.

The HMBC correlations (Fig. 2) from H-2 ( $\delta_H$  3.95) to C-1 ( $\delta_C$  41.1), C-3 ( $\delta_C$  198.7), and C-11 ( $\delta_C$  20.8), the COSY correlation of H<sub>3</sub>-10 ( $\delta_H$  1.12) with H-9 ( $\delta_H$  4.14) and HMBC correlation of H-7 ( $\delta_H$  5.59) with C-9 ( $\delta_C$  66.2), respectively, allowed two hydroxyl groups to be located at C-2 and C-9, respectively and carbonyl function at C-3. The cross peaks between H<sub>3</sub>-13 ( $\delta_H$  1.85) with C-4 ( $\delta_C$  123.7), C-5 ( $\delta_C$  162.0), and C-6 ( $\delta_C$  56.1) assigned the position of CH<sub>3</sub>-13 and an enone system.

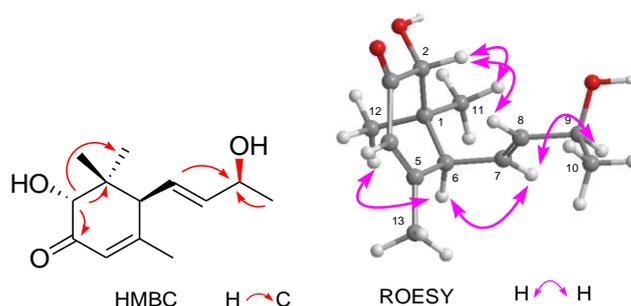
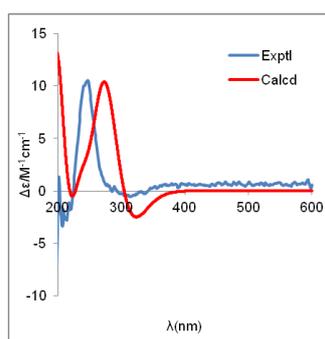


Figure 2. Key HMBC, and ROESY correlations of compound 1.

The relative configuration in **1** was determined by the ROESY experiment (Fig. 2). The observed ROESY correlations: H-2 with H-8 and H-11, H-6 with H-7 and H-12, and H-7 with H-9 suggested that HO-2 and H-6 were of the same orientation on the  $\alpha$ -face of the six-membered ring, HO-9 was the  $\beta$ -oriented, and the double bond was *E* configuration. Moreover, the amount of coupling constant ( $J = 15.2$  Hz) between H-7 and H-8 substantiated *E* configuration of the double bond. In order to determine the absolute configuration of C-6 and C-9 the experimental and calculated ECD spectra predicted by TDDFT (Fig. 3) were carried out.



**Figure 3.** Calculated and experimental ECDs of **1** (red, calculated at the B3LYP-PCM/6-31G(d,p)//B3LYP/6-31G(d,p) level in CH<sub>3</sub>OH; blue, experimental in CH<sub>3</sub>OH).

The results (Fig. S2 and S6) revealed the structures of 1-A and 1-C were in accordance with the experimental spectra. Further comparing experimental and calculated <sup>13</sup>C NMR data (Fig. S22 and 23) gave the *S*-configuration of chiral center at C-9 ( $\Delta\delta$  3.3) rather than *R*-configuration ( $\Delta\delta$  4.6). Finally, the structure of compound **1** was assigned as (2*R*,6*S*,9*S*)-2-hydroxy-3-oxo- $\alpha$ -ionol.

The known compounds were identified to be (6*S*,9*S*)-6-hydroxy-3-oxo- $\alpha$ -ionol (**2**) [17], (6*R*,9*S*)-3-oxo- $\alpha$ -ionol  $\beta$ -D-glucopyranoside (**3**) [18] and (6*R*,9*R*)-3-oxo- $\alpha$ -ionol  $\beta$ -D-glucopyranoside (**4**) [18] by comparing their spectroscopic data with those in the literature.

**Table 1.** NMR data for compounds **1–4** (TMS as the internal standard,  $\delta$  in ppm)<sup>a</sup>

No.	<b>1</b> <sup>a,c</sup>		<b>2</b> <sup>b,c</sup>	<b>3</b> <sup>a,d</sup>	<b>4</b> <sup>a,d</sup>
	$\delta_{\text{H}}$ <i>J</i> (Hz)	$\delta_{\text{C}}$	$\delta_{\text{C}}$	$\delta_{\text{C}}$	$\delta_{\text{C}}$
1		41.1 (s)	40.9 (s)	37.2 (s)	37.1 (s)
2	3.95 (1H, br s)	75.5 (d)	49.4 (t)	48.7 (t)	48.3 (t)
3		198.7 (s)	197.4 (s)	202.1 (s)	202.1 (s)
4	5.84 (1H, s)	123.7 (d)	125.5 (d)	126.4 (d)	126.1 (d)
5		162.0 (s)	164.4 (s)	165.7 (s)	165.9 (s)
6	2.67 (1H, br d, <i>J</i> = 8.6 Hz)	56.1 (d)	77.8 (s)	57.0 (d)	56.8 (d)
7	5.59 (1H, dd, <i>J</i> = 15.2, 8.6 Hz)	124.3 (d)	127.9 (d)	131.6 (d)	128.8 (d)
8	5.63 (1H, dd, <i>J</i> = 15.2, 5.1 Hz)	139.8 (d)	135.9 (d)	137.2 (d)	138.2 (d)
9	4.14 (1H, m)	66.2 (d)	66.1 (d)	74.9 (d)	77.0 (d)
10	1.12 (3H, d, <i>J</i> = 4.2 Hz)	23.9 (q)	24.1 (q)	22.3 (q)	21.0 (q)
11	0.81 (3H, s)	20.8 (q)	24.0 (q)	27.9 (q)	27.6 (q)
12	0.96 (3H, s)	24.6 (q)	23.1 (q)	28.2 (q)	28.0 (q)
13	1.85 (3H, s)	22.9 (q)	19.0 (q)	23.9 (q)	23.8 (q)
1'				101.1 (d)	102.4 (d)
2''				75.1 (d)	75.3 (d)
3''				78.4 (d)	78.1 (d)
4'				71.8 (d)	71.5 (d)
5'				78.4 (d)	78.0 (d)
6'				63.0 (t)	62.7 (t)

<sup>a</sup> NMR data were recorded at 600 MHz, <sup>b</sup> NMR data were recorded at 400 MHz,

<sup>c</sup> NMR data were recorded in DMSO-*d*<sub>6</sub>, <sup>d</sup> NMR data were recorded in CD<sub>3</sub>OD.

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## Supporting Information

Supporting Information accompanies this paper on <http://www.acgpubs.org/RNP>

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